

Available online at http://www.journalcra.com

International Journal of Current Research Vol. 3, Issue, 6, pp.154-159, June, 2011 INTERNATIONAL JOURNAL OF CURRENT RESEARCH

### **RESEARCH ARTICLE**

# ENVIRONMENTAL ASSESSMENT OF PETROCHEMICAL WASTEWATER DISCHARGE INTO ELEME RIVER: A CASE STUDY OF PETROCHEMICAL INDUSTRY IN NIGERIA

## Ukpaka, C. P.

Department of Chemical/Petrochemical Engineering, Rivers State University of Science and Technology, Nkpolu, P.M.B. 5080, Port Harcourt, Nigeria

### ARTICLE INFO

Article History: Received 1<sup>st</sup> February, 2011 Received in revised form 5<sup>th</sup> March, 2011 Accepted 9<sup>th</sup> May, 2011 Published online 2<sup>nd</sup> June 2011

*Key Words:* Environmental,

Assessment, Petrochemical wastewater, River.

**INTRODUCTION** 

### ABSTRACT

The study was based on environmental assessment of petrochemical wastewater parameters discharged to Eleme River. Three methods: physical, chemical and biological were employed for the assessment on the physiochemical parameters. Laboratory tests were done on the sample using FEPA set standard for industrial effluent water test. The data obtained from the laboratory tests, further assessment was performed to examine the impact of these parameters to the receiving water body with respect to FEPA standard for industrial wastewater discharge to the inland water. The result showed that some parameters; Biochemical Oxygen Demand (BOD), Total Suspended Solids (TSS), pH, Temperature and Total Dissolved Solid (TDS), TSS, pH, Temperature, and TDS of the wastewater exceeded the FEPA standard by 80%, 91%, 76%, 24%, and 71% respectively. The conceptual model used for the existing treatment plant in Eleme Petrochemical is inadequate and needs improvement. The paper presented a new model as a way forward to improve the quality of wastewater generate from Eleme Petrochemical. Therefore, a new treatment plant design calculations was carried out to modify the existing plant for optimum performance. Finally, the information obtained from this study should be useful tools for further assessment of petrochemical wastewater parameters even to other industrial wastewater in nature.

© Copy Right, IJCR, 2011 Academic Journals. All rights reserved

The modern civilization accompanied by the rapid growth of population increase and in hydrocarbon utilization industrialization has created the situation of huge amount of production of wastewater per day. All over the world today the industries are vexed with the problem of wastewater disposal and management of generated wastewater. Laboratory studies were conducted to assess the petrochemical wastewater generated which was discharged to Eleme River in Niger Delta Area of Nigeria. This study reports on the concentration of physiochemical parameters of the wastewater in Eleme River. Investigation conducted by various research groups revealed that Global fisheries is facing constant decline in the fish stocks, both in coastal and in land water pollution (EG, 1991; Gevard, 1999; James, 1980; King, 1978; Klyhn, 1979; Lynch and Poole, 1999 and Nwaigazie, 2000). The decline can be attributed to constant discharge of effluent water into the lake, river, sea etc. The physiochemical parameters are of the major adversely influencing most body functions. Exposure to Cd is known to produce symptoms like abdominal cramps, nausea, salivation, diarrhea, hypertension, liver Girrhosis etc in human. It may sometimes prove to be carcinogenic or produce teratogenic effects. Thus for healthy fish production, it is very important to evaluate the harmful effects of physiochemical parameters in combination. In the present studies, an attempt has been made to study the effects of physiochemical parameters in water body of fresh and salt water media. (Raven, Berg and Johnson, 1998; Marshal, 1973; Onuoha,

\*Corresponding author: chukwuemeka24@yahoo.com

1994 and Keep 1998, Ukpaka, 2004, 2005, and 2006). The organisms living in water environment needs good water for building up body tissue, which further signifies that a balanced diet is necessary for proper functioning of the body. Recent understanding of different biochemical processes has proved useful in determining the mechanisms of toxicity in different protective mechanisms of the body to combat the toxic effect of the pollutants, besides it is also observed that some biochemical alteration occurring in the body give the first indication of stress in the organism and hence the efforts, on the part of pollution, biologists to explore the possibility of making use of the phenomenon to locate certain type of pollutants on nature. The assessment carried out on Eleme River revealed that small amount of many relative toxic materials released into an aquatic environment cause multiple change in the internal dynamics of aquatic organism in different levels (Benfled and Bandall, 1980; Davis and Cornwell, 19999; Hurley, 1989; Handan, 1976; King, 1978; Atlas, 1988; AWWA, 1990; Camp and Messerve, 1974; Barta, 1988; Chikere, 2000 and Eckenfelder, 1989). The physiochemical parameters of interest includes; temperature, pH, conductivity, turbidity, Dissolved Oxygen (DO), Total Suspended Solids (TSS), alkalinity, Biochemical Oxygen Demand (BOD), oil and grease, phenol, chloride, phosphate, nitrate, sulphate, iron, cadmium, lead, sodium, copper and manganese. From the available literature it appears that the study of physiochemical parameters are very useful in understanding the mechanism of its effect on the water body (Ukpaka, 2007, 2008, 2009 and 2010); PPAGPMI, 2001; Khan, Shaika, Zarina, Lanjekar and Nayab, 2007; Martin and Kulathooran, 2007 and Snnikenthan, Shrihari and Pradecpan, 2007). The present investigation has been conducted to determine the quality of Elele River with respect to the physicochemical parameters.

### **MATERIALS AND METHOD**

The liquid waste, which is generated by Eleme Petrochemical complex are normally discharge into the Eleme River the South of the complex. The Eleme River emptied into Bonny River through Okpokoro River at Okirika arm for final The Eleme Petrochemical Industry (EPI) disposal. wastewaters were used for the case study. Four sampling points of wastewater were collected within the petrochemical complex namely: sample point 1 at equalization basin, sample point 2 at final discharge basin, about 2.5m away from the equalization basin. This basin is the point where all treatment effluent water from petrochemical complex are pumped to, before finally discharged into the Eleme River, sample point 3 at dissolve air floatation basin, where chemical treatment takes place to coagulate the suspended particles into sludge and scraped before, sending it into biological treatment used for this study were collected with 1litre plastic cans previously washed thoroughly with treated potable water and dried, before use, also they were rinsed with the sample before finally collection of the sample. The interval of collection was once every month, and the containers were labeled accordingly. All samples were transported to the petrochemical complex laboratory and analysed immediately.

#### Laboratory Tests

Different laboratory test wee conducted using the APHA, (1989) method.

#### The Model

#### **Equalization Basin Design**

The major function of equalization basin is to minimize or control fluctuation in wastewater characteristics in order to provide optimum conditions for subsequent treatment processes. For petrochemical industry wastewater, the hourly flow used for design is shown on Table 1 and calculation is based on material balance produce as stated by Patterson and Menez adopted from Eckenfelder, (1989).

(i) Volume of equalization is given as;

- V = average flow x retention time. (1)
- (ii) Effluent concentration is calculated as follows using material balance for equalization basin as presented by Eckenfelder, (1989).

$$C_1QT + C_0V = C_2QT + C_2V$$
(1a)

Making  $C_2$  the subject of the formula, equation (1a) becomes;

$$C_{2} = \frac{C_{1}t + C_{o}V/Q}{T + V/Q}$$
(2)

#### **Biological Aeration Tank Design (Activated Sludge)**

Completely Mixed Activated Sludge Process Design

The design formulas are by mass balance application and the source of design formula is adopted from introduction to environmental Engineering (Davis and Cornwell, 1998). For steady state condition mass balance is stated as;

 $\begin{bmatrix} Biomass + biomass \\ Influent \ accumulates \end{bmatrix} = \begin{bmatrix} Biomass + biomass \\ Effluent \ wasted \end{bmatrix}$ (3)

$$QX_{0} + \left[V\frac{\mu mSX}{KS+S}\right] - k_{d}X = (Q - Q_{w})X_{e} + Q_{w}X_{r}$$
<sup>(4)</sup>

Also in designing aeration tank, the mass balance equation for soluble  $BOD_5$  (food for microorganism) in steady stated is written as

Food = food consumed-food in

In influent by microorganisms effluent wasted (5) By formula;

$$QS_o - (V)\frac{\mu_m SX}{Y(K_s + S)} = (Q - Q_w S + Q_w S)$$
<sup>(6)</sup>

Supposing percent standard FEPA standard allowable solid by the influent. Therefore calculating allowable suspended solid(s):

 $S = Total BOD_5 Allowed - BOD_5$  in suspended solid.

Also the residence time is calculated with the formula:  

$$S = \frac{KS \left(1 + K_{d} \theta_{c}\right)}{\theta_{c} (\mu_{m} - K_{d}) - 1}$$
(7)

where 
$$\theta_c = \frac{VX}{Q_w X_r}$$
  
:.  $132 = \frac{100 \ \theta[(1 = 0.050)\theta_c]}{\theta_c(2.5 - 0.050) - 1}$ 

For calculation of detention time (t), the formula stated below is useful

$$X = \theta_c \frac{(Y)(S_o - S)}{\theta_c (1 + K_d O_c)}$$
(8)

#### **Chemical Treatment Design**

#### Flocculation Tank Design

The important factor is to bring the particles into contact so that they collide and stick together and grow to size that will readily settle component design Baffled tank design.

For rapid mixing tank, the design formula is

$$P = KT (n)^{3} (D)^{5} \rho$$
Drag force on the paddle is given as
(9)

$$P = \frac{CDA\rho(V\rho)^3}{2} \tag{10}$$

Design equation is

$$T_{o} = \frac{V}{Q}$$
(11)

Therefore, the equation by Eckenfelder, (1989) for mixing volume is given as;

$$G = \sqrt{\frac{P}{\mu V}}$$
(12)

$$\frac{W}{G^2\mu} = P$$

Using the equation

Di = 
$$\frac{(P)^{15}}{[(\rho K_T)(n)(\rho)]}$$

For peak factor (PF) Determination: therefore, the peak factor of the flow is defined by Davis and Cornwell, (1998) as;

$$Peak \ flow = \frac{\text{peak factor} \times \text{influent} \times flow \ rate}{Overflow \ rate \left(m^3 / m^2\right)}$$
(13)

#### **Sedimentation Basin Design**

In designing discrete setting, basic factor affecting setting is the terminal velocity and it is defined by the equation

$$V = \sqrt{\frac{4g(\rho s - \rho l)D}{3C_d \rho}}$$
(14)

Reynolds number less than 1.0 viscous force predominate and the formula becomes;

$$C_d = \frac{24}{NR_e}$$
 and  $NR_e = \frac{\rho S - \rho l}{\mu}$ 

where  $\mu$  is the viscosity of the liquid substituting into equation (6) becomes;

$$C_d = \frac{\rho S - \rho l}{18\mu} g D^2 \tag{15}$$

### **RESULTS AND DISCUSSION**

Parameters of interest were studied on investigating the physiochemical component of effluent water discharge into Eleme River. The following physiochemical parameters were analysed: pH, temperature, conductivity, turbidity, Dissolved Oxygen (DO), Total Suspended Solid (TSS), Alkalinity, Total Dissolved Solid, (TDS), Biochemical Oxygen Demand (BOD), phenol, chloride, phosphate, nitrate, sulphate, iron, cadmium, lead, sodium, copper and manganese. The results of different sampling point are present in tables for the various month of investigation. The variation in the physiochemical parameters concentration can be attributed to constant discharge of wastewater in the receiving body (Eleme River) as well as other environmental factors. Similarly, the variation in the physiochemical parameters concentration can be attributed to the following: inadequate treatment in the sedimentation basin unit, chemical treatment unit, activated sludge unit, and equalization basin unit as well as the flow characteristics of the Eleme River. The summary of the results are present in Table 1 to 8 for selected physiochemical parameter.

The results of the analysis obtained from one month indicate increase in some physicochemical parameters in different sampling points as well decrease in concentration of the parameters as shown in Table 1. Some of the parameters concentrations are within the international acceptable limit while some are above. This means that some of the physicochemical parameters concentration, its characteristics influence the Eleme River as a result of effluent discharge. The results showed above 80% increase in BOD concentration, 91% on TSS concentration, 76% on pH concentration, 24% on temperature and 71% on TDS concentration whereas other parameters are within the acceptable limit. From Table 2, it is seen that decrease in physiochemical parameters are obtained as the samples are analysed from one point to the other as well increase as presented. Increase in physicochemical parameters can be attributed to continuous discharge of effluent water on the river as well as constant accumulation of contaminants. Similarly, the increase in the physiochemical parameters can be attributed to low dissolution rate, high rate of spreading, poor treatment method applied, high volume of wastewater discharge into the river etc. The result of physicochemical parameters for 3months sampling in Eleme River indicate increase in some physicochemical parameters above the acceptable limit as recommended by International Bodies. Similarly, some of the physicochemical parameters are within the acceptable limit. Parameters of temperature, pH, BOD, TDS, and TSS concentration are above the acceptable limit with values of 24%, 76%, 80%, 71% and 91% respectively. The changes in the physicochemical parameters of Eleme River can be attributed to continuous discharge of wastewater, high quality of wastewater discharged, low spreading rate of wastewater, and other environmental factors as shown in Table 3.

In Table 4 results on the physicochemical parameters continued to show increase in contamination level of Eleme River with increase in continuous discharge of wastewater. The analysis results obtained for 4 months. Sampling indicate 24%, 76%, 80%, 71% and 91% increase in temperature, pH, BOD, TDS and TSS. These parameters concentration are above the acceptable limit as recommended by the international Bodies. The continuous increase in the physicochemical parameters can be attributed to the plant design factor such as equalization basin design, biological aeration tank design, chemical treatment design, sedimentation basin design, and experimental test. The results of the analysis obtained from one month indicate increase in some physicochemical parameters in different sampling points as well decrease in concentration of the parameters as shown in Table 5. Some of the parameters concentrations are within the international acceptable limit while some are above. This means that some of the physicochemical parameters concentration, its characteristics influence the Eleme River as a result of effluent discharge. The results showed above 80% increase in BOD concentration, 91% on TSS concentration, 76% on pH concentration, 24% on temperature and 71% on TDS concentration whereas other parameters are within the acceptable limit.

Results obtained in Table 2, illustrates the characteristics of the physicochemical parameters. From Table 2, it is seen that decrease in physiochemical parameters are obtained as the samples are analysed from one point to the other as well increase as presented. The increase in physicochemical parameters can be attributed to continuous discharge of effluent water on the river as well as constant accumulation of contaminants. Similarly, the variation in the physiochemical parameters can be attributed to low dissolution rate, high rate of spreading, poor treatment method applied, high volume of wastewater discharge into the river etc. The result of physicochemical parameters for 7 months sampling in Eleme River indicate increase in some physicochemical parameters above the acceptable limit as recommended by International Bodies. Similarly, some of the physicochemical parameters are within the acceptable limit. Parameters of temperature, pH, BOD, TDS, and TSS concentration are above the acceptable limit with values of 24%

Table 1: Result of physiochemical parameters for month 1

| Parameters                                 | SP.01   | SP.02   | SP.03   | SP.04   |
|--|---------|---------|---------|---------|
| Temperature °C                             | 24.7    | 26.4    | 25.4    | 27      |
| pH   | 6.90    | 7.33    | 5.85    | 5.99    |
| Cond µ/cm                                  | 996     | 329     | 352     | 76.20   |
| Turbidity NTU                              | 10.96   | 8.28    | 37.23   | 3.9     |
| DO mg/l                                    | 6.5     | 7.43    | 5.23    | 3.9     |
| TSS mg/l                                   | 118.45  | 20.10   | 8.11    | 6.85    |
| Alkalinity mg/l                            | 60.15   | 17      | 47      | 26      |
| TDS mg/l                                   | 468.0   | 153.8   | 24.5    | 7.89    |
| BOD <sub>5</sub> mg/l                      | 2.4     | 1.00    | 2.00    | 1.5     |
| Phenol mg/l                                | 0.26    | < 0.001 | 0.07    | < 0.01  |
| Chloride mg/l                              | 12.02   | 0.48    | 5.01    | 0.93    |
| Phosphate mg/l                             | 2.87    | <0.1    | 1.65    | <0.1    |
| Nitrate mg/l                               | 6.62    | 4.16    | 5.8     | 4.32    |
| Sulphate mg/l                              | 0.760   | 0.037   | 0.511   | 0.73    |
| Iron mg/l                                  | 0.233   | 0.058   | 0.129   | 0.120   |
| Cadmium mg/l                               | 0.0233  | 0.004   | 0.013   | 0.008   |
| Lead mg/l                                  | < 0.001 | < 0.001 | < 0.001 | < 0.001 |
| Sodium mg/l                                | 7.012   | 50.896  | 5.581   | 3.428   |
| Copper mg/l                                | 0.09    | 0.004   | 0.029   | 0.012   |
| Manganese mg/l                             | < 0.001 | < 0.001 | < 0.001 | < 0.001 |
| NB: SP = Sampling Point: 01, 02, 03 and 04 |         |         |         |         |

Table 2: Result of physiochemical parameters for month 2

| Parameters            | SP.01   | SP.02   | SP.03   | SP.04   |
|-----------------------|---------|---------|---------|---------|
| Temperature °C        | 26.1    | 25.2    | 25.4    | 24.8    |
| pH                    | 8.3     | 7.05    | 5.55    | 6.19    |
| Cond µ/cm             | 927     | 140     | 924     | 135     |
| Turbidity NTU         | 47.69   | 2.38    | 13.83   | 2.95    |
| DO mg/l               | 3.8     | 6.94    | 4.6     | 5.8     |
| TSS mg/l              | 96.75   | 4.68    | 70.98   | 30.85   |
| Alkalinity mg/l       | 230     | 24      | 90      | 48      |
| TDS mg/l              | 449.36  | 428.7   | 11.9    | 58.93   |
| BOD <sub>5</sub> mg/l | 2.8     | 1.01    | 2.2     | 195     |
| Oil and Grease mg/l   | 11.3    | 1.54    | 5.8     | 2.4     |
| Phenol mg/l           | 0.23    | < 0.001 | 0.06    | < 0.01  |
| Chloride mg/l         | 17.02   | 1.55    | 13.01   | 3.93    |
| Phosphate mg/l        | < 0.001 | < 0.001 | 0.001   | 0.001   |
| Nitrate mg/l          | 4.06    | 24.16   | 3.8     | 3.13    |
| Sulphate mg/l         | 0.600   | 0.030   | 0.556   | 0.185   |
| Iron mg/l             | 0.233   | 0.058   | 0.129   | 0.120   |
| Cadmium mg/l          | 0.001   | < 0.001 | 0.007   | 0.003   |
| Lead mg/l             | < 0.001 | < 0.001 | < 0.001 | < 0.001 |
| Sodium mg/l           | 7.012   | 50.896  | 5.581   | 3.428   |
| Copper mg/l           | 0.09    | 0.004   | 0.029   | 0.012   |
| Manganese mg/l        | < 0.001 | < 0.001 | < 0.001 | < 0.001 |

NB: SP = Sampling Point: 01, 02, 03 and 04

Table 3: Result of Physiochemical Parameters for month 3

|                       | 67 A 4  | <b>GD 00</b> | <b>CD 00</b> | <b>GD 04</b> |
|-----------------------|---------|--------------|--------------|--------------|
| Parameters            | SP.01   | SP.02        | SP.03        | SP.04        |
| Temperature °C        | 27.23   | 27.06        | 26.45        | 26.15        |
| pH                    | 7.09    | 6.73         | 5.67         | 6.73         |
| Cond µ/cm             | 85      | 287          | 129.4        | 168.6        |
| Turbidity NTU         | 13.22   | 3.47         | 11.07        | 6.79         |
| DO mg/l               | 4.33    | 7.32         | 2.90         | 3.88         |
| TSS mg/l              | 3.89    | 4.95         | 3.02         | 2.9          |
| Alkalinity mg/l       | 57      | 38           | 25           | 24           |
| TDS mg/l              | 5.63    | 134          | 68.02        | 98.11        |
| BOD <sub>5</sub> mg/l | 2.8     | 1.00         | 2.6          | 2.0          |
| Oil and Grease mg/l   | 8.5     | 0.8          | 5.4          | 1.9          |
| Phenol mg/l           | 0.36    | < 0.001      | 0.067        | 0.03         |
| Chloride mg/l         | 29.12   | 27.07        | 26.54        | 21.00        |
| Phosphate mg/l        | 8.00    | 0.47         | 3.3          | 0.58         |
| Nitrate mg/l          | 4.14    | 6.61         | 4.02         | 1.75         |
| Sulphate mg/l         | 1.250   | 0.187        | 0.42         | 0.216        |
| Iron mg/l             | 2.65    | 0.68         | 0.089        | 0.04         |
| Cadmium mg/l          | 0.03    | < 0.001      | 0.007        | 0.003        |
| Lead mg/l             | 0.134   | 0.132        | 0.046        | < 0.012      |
| Sodium mg/l           | 4.35    | 7.53         | 5.22         | 4.04         |
| Copper mg/l           | 0.015   | 0.002        | 0.012        | 0.005        |
| Manganese mg/l        | < 0.001 | < 0.001      | < 0.001      | < 0.001      |

NB: SP = Sampling Point: 01, 02, 03 and 04

76%, 80%, 71% and 91% respectively. The changes in the physicochemical parameters of Eleme River can be attributed to continuous discharge of wastewater, high quality of wastewater discharged, low spreading rate of wastewater, and other environmental factors as shown in Table 7. In Table 8 results on the physicochemical parameters continued to show increase in

#### Table 4: Result of Physiochemical Parameters for month 4

| Parameters            | SP.01   | SP.02   | SP.03         | SP.04   |
|-----------------------|---------|---------|---------------|---------|
| Temperature °C        | 24.69   | 24.88   | 25.45         | 24.93   |
| pH                    | 6.40    | 6.78    | 5.97          | 6.12    |
| Cond µ/cm             | 66.7    | 269     | 124.14        | 159.4   |
| Turbidity NTU         | 11.52   | 1.98    | 9.7           | 5.66    |
| DO mg/l               | 4.77    | 5.3     | 5.96          | 5.0     |
| TSS mg/l              | 20.97   | 7.30    | 18.79         | 12.13   |
| Alkalinity mg/l       | 34      | 18      | 25.4          | 29      |
| TDS mg/l              | 32      | 128     | 59            | 76      |
| BOD <sub>5</sub> mg/l | 2.8     | 1.09    | 2.2           | 2.0     |
| Oil and Grease mg/l   | 10.25   | 1.8     | 5.8           | 2.3     |
| Phenol mg/l           | 0.425   | 0.302   | 0.754         | 0.349   |
| Chloride mg/l         | 5.29    | 27.36   | 36.36         | 29.00   |
| Phosphate mg/l        | 2.84    | <1.067  | 0.605         | 4.78    |
| Nitrate mg/l          | < 0.813 | <1.067  | 6.45          | 4.78    |
| Sulphate mg/l         | 0.813   | 0.039   | 0.605         | 0.193   |
| Iron mg/l             | 0.169   | 0.083   | 0.101         | 0.05    |
| Cadmium mg/l          | 0.024   | 0.035   | 0.032         | 0.016   |
| Lead mg/l             | 0.153   | 0.152   | 0.059         | 0.072   |
| Sodium mg/l           | 4.35    | 7.53    | 5.22          | 4.04    |
| Copper mg/l           | 0.001   | 0.006   | Insignificant | 0.003   |
| Manganese mg/l        | < 0.001 | < 0.001 | < 0.001       | < 0.001 |

NB: SP = Sampling Point: 01, 02, 03 and 04

Table 5: Result of Physiochemical Parameters for month 5

| Parameters            | SP.01         | SP.02     | SP.03   | SP.04   |
|-----------------------|---------------|-----------|---------|---------|
| Temperature °C        | 29            | 28.5      | 29.35   | 27      |
| pH                    | 9.10          | 7.5       | 6.9     | 7       |
| Cond µ/cm             | 197           | 309       | 108     | 242     |
| Turbidity NTU         | 179           | 7.9       | 300     | 71      |
| DO mg/l               | 4.7           | 5.7       | 3.9     | 3       |
| TSS mg/l              | 90.5          | 4.59      | 33      | 27      |
| Alkalinity mg/l       | 20            | 13        | 22      | 15      |
| TDS mg/l              | 66            | 198       | 48      | 23      |
| BOD <sub>5</sub> mg/l | 2.3           | 0.7       | 1.9     | 1.1     |
| Oil and Grease mg/l   | 7.0           | 1.2       | 2.1     | 1.6     |
| Phenol mg/l           | 0.42          | 0.01      | 0.19    | 0.03    |
| Chloride mg/l         | 4.56          | 10.36     | 2.02    | 1.97    |
| Phosphate mg/l        | 2.26          | 0.06      | 2.02    | 0.72    |
| Nitrate mg/l          | 4.45          | 2.64      | 3.92    | 2.2     |
| Sulphate mg/l         | 4.48          | 0.87      | 3.03    | 2.05    |
| Iron mg/l             | 0.197         | 0.073     | 0.1004  | 0.039   |
| Cadmium mg/l          | 0.024         | 0.035     | 0.032   | 0.016   |
| Lead mg/l             | 0.169         | 0.167     | 0.063   | 0.79    |
| Sodium mg/l           | 8.95          | 11.05     | 1.32    | 0.982   |
| Copper mg/l           | 0.003         | < 0.001   | < 0.081 | < 0.001 |
| Manganese mg/l        | < 0.001       | < 0.001   | < 0.001 | < 0.001 |
| NB: SP = Sampling Poi | nt: 01, 02, 0 | 03 and 04 |         |         |

Table 6: Result of Physiochemical Parameters for month 6

| Parameters            | SP.01  | SP.02   | SP.03  | SP.04 |
|-----------------------|--------|---------|--------|-------|
| Temperature °C        | 30     | 28      | 31     | 30    |
| pH                    | 6.58   | 12.90   | 12.90  | 9.25  |
| Cond µ/cm             | 78.5   | 6.52    | 62.23  | 45.65 |
| Turbidity NTU         | 179    | 7.9     | 101.52 | 210   |
| DO mg/l               | 4.5    | 5.0     | 12.54  | 10.5  |
| TSS mg/l              | 90.5   | 4.59    | 100    | 101   |
| Alkalinity mg/l       | 20     | 13      | 55     | 65    |
| TDS mg/l              | 66     | 176     | 150    | 120   |
| BOD <sub>5</sub> mg/l | 3.8    | 1.3     | 5.12   | 2.85  |
| Oil and Grease mg/l   | 9.0    | 3.2     | 2.54   | 1.54  |
| Phenol mg/l           | 0.42   | 0.01    | 0.65   | 0.54  |
| Chloride mg/l         | 6.10   | 12.430  | 12.15  | 15    |
| Phosphate mg/l        | 2.3    | 1.9     | 15     | 3.52  |
| Nitrate mg/l          | 4.45   | 2.64    | 3.565  | 5.23  |
| Sulphate mg/l         | 5.09   | 272.7   | 120    | 201   |
| Iron mg/l             | 0.079  | 0.064   | 0.21   | 0.056 |
| Cadmium mg/l          | 0.055  | 0.17    | 0.015  | 0.045 |
| Lead mg/l             | -0.036 | -0.030  | 0.045  | 0.012 |
| Sodium mg/l           | 7.89   | 12.06   | 5.62   | 2.01  |
| Copper mg/l           | 0.003  | < 0.001 | -0.25  | 0.012 |
| Manganese mg/l        | 0.051  | 0.032   | 0.081  | 0.021 |

contamination level of Eleme River with increase in continuous discharge of wastewater. The analysis results obtained for 4 months. Sampling indicate 24%, 76%, 80%, 71% and 91% increase in temperature, pH, BOD, TDS and TSS. These parameters concentration are above the acceptable limit as recommended by the international Bodies. The continuous

Table 7: Result of Physiochemical Parameters for month 7

| 31<br>9.89<br>172 | 39<br>12.09  | 29<br>7.07   | 29.8   |
|-------------------|--|--|--|
| 9.89<br>172       | 12.09  | 7.07   |  |
| 172               |  | 1.01   | 6.64   |
|                   | 79.03  | 230  | 69.8   |
| 156               | 9.23   | 101  | 150  |
| 3.8               | 4.4  | 3.4  | 2.3  |
| 123.6             | 6.86   | 15   | 25   |
| 22                | 9.86   | 5.23   | 5.2  |
| 66                | 176  | 81   | 96   |
| 4.21              | 2.58   | 25   | 29   |
| 9.0               | 3.2  | 1.2  | 12   |
| 0.56              | 0.01   | 0.012  | 0.06   |
| 5.002             | 27.512   | 18.52  | 0.6789   |
| 1.5               | 1.5  | 4.50   | 0.5  |
| 3.506             | 4.023  | 2.00   | 0.42   |
| 0.139             | 3.090  | 0.116  | 0.048  |
| 0.017             | 0.064  | 0.011  | 0.03   |
| 0.0102            | -0.017   | -0.0120  | -0.0010  |
| 0.036             | -0.030   | 0.005  | -0.0010  |
| 9.804             | 11.86  | 0.26   | 9.02   |
| 0.006             | < 0.003  | < 0.0012   | 0.004  |
| 0.010             | < 0.003  | 0.054  | 0.021  |
|                   | 156<br>3.8<br>123.6<br>22<br>66<br>4.21<br>9.0<br>0.56<br>5.002<br>1.5<br>3.506<br>0.139<br>0.017<br>0.0102<br>0.036<br>9.804<br>0.006<br>0.0010 | $\begin{array}{cccccccccccccccccccccccccccccccccccc$ | $\begin{array}{cccccccccccccccccccccccccccccccccccc$ |

NS: Sr = Sampling Folic, 01, 02, 05 and 04

| Table 8: Result of Physioch | emical Parameters f | for month 8 |
|-----------------------------|---------------------|-------------|
|-----------------------------|---------------------|-------------|

| Parameters                                | SP.01  | SP.02    | SP.03     | SP.04    |
|---|--------|----------|-----------|----------|
| Temperature °C                            | 30     | 31       | 31        | 29       |
| pН  | 7.00   | 9.89     | 12.90     | 11.00    |
| Cond µ/cm                                 | 79     | 9.25     | 148       | 98       |
| Turbidity NTU                             | 109    | 10.12    | 5.2       | 98       |
| DO mg/l                                   | 4.03   | 5.4      | 2.02      | 9.2      |
| TSS mg/l                                  | 81.9   | 4.96     | 26.9      | 222      |
| Alkalinity mg/l                           | 20     | 13       | 5.02      | 8.3      |
| TDS mg/l                                  | 128    | 141      | 5.9       | 26       |
| BOD <sub>5</sub> mg/l                     | 2.809  | 2.39     | 1.01      | 0.59     |
| Oil and Grease mg/l                       | 8.95   | 2.7      | 1.2       | 3.3      |
| Phenol mg/l                               | 0.42   | 0.01     | 0.02      | 0.06     |
| Chloride mg/l                             | 11.56  | 14.00    | 3.9       | 6.2      |
| Phosphate mg/l                            | 1.03   | 1.23     | -0.25     | 0.78     |
| Nitrate mg/l                              | 5.02   | 4.97     | 4.2       | 6.13     |
| Sulphate mg/l                             | 5.09   | 272.7    | 7.32      | 4.13     |
| Iron mg/l                                 | 0.080  | 0.043    | 0.005     | 0.07     |
| Cadmium mg/l                              | 0.070  | 0.025    | 0.015     | 0.075    |
| Lead mg/l                                 | -0.029 | 0005     | -0.001    | -0.001   |
| Sodium mg/l                               | 7.89   | 12.06    | 5.12      | 9.10     |
| Copper mg/l                               | 0.003  | < 0.0021 | -0.014    | -0.045   |
| Manganese mg/l                            | <0010  | < 0.003  | < 0.00101 | < 0.0054 |
| B: SP = Sampling Point: 01, 02, 03 and 04 |        |          |           |          |

increase in the physicochemical parameters can be attributed to the plant design factor such as equalization, basin design, biological aeration tank design, chemical treatment design, sedimentation basin design, and experimental test.

#### Conclusion

The present study revealed an interesting insight on a few physiochemical parameters of the wastewater discharge in Eleme River. The results showed an increase in Biochemical Oxygen Demand (BOD) value of 80%, Total Suspended Solids (TS) value of 91%, pH value of 76%, temperature value of 24% and Total Dissolved Solids (TDS) value of 71%. Depending upon the source, nature quality and quantity of wastewater generation, the variation occurs in the experimental value of the analysed parameters for the different samples. This study gives an idea about the quantitative aspect and nature of the wastewater discharge into the Eleme

River. It can be expected that this study will inspire further investigation regarding the other aspects and characteristics of the physicochemical parameters of the wastewater after eight months.

The following conclusion was drawn from the study:

- 1. The composition of wastewater discharge into the Eleme River by the petrochemical Industry is hazardous and the characteristics shows that the chemicals composition is above the acceptable limit as recommended by International bodies.
- 2. The treatment techniques put in place by the company is inadequate to handle the wastewater generated per day.
- 3. The treatment equipment in place should be replaced with new one of modern technology.
- 4. Constant treatments on the Eleme river should be carried out in every 3months interval as a way forward to enhance environmental clean up as well as increase the portability of River water for domestic use, aquatic use etc.
- 5. If measures are not taken by the Federal Government to stop this habit exacted by Oil companies in Nigeria in terms of waste treatment and management there is tendency that the ground water within the vicinity of Eleme with contaminated in due cost.
- 6. This paper present the current status of Eleme River in terms of physiochemical parameters. One may say that the level or the degree of pollutant in the Eleme River is majorly cost by the industrial activities taken place with the area. Although there are domestic waste disposed in the Eleme River but the impact is not significant, because the microorganisms present in the aquatic environment needs them for then food.

Therefore, this paper recommend adequate treatment of wastewater as well ensure that hazardous components are not discharge into the Eleme River, since the River save as a major source for drinking, domestic uses, industrial uses etc.

### Nomenclature

| G                | = | Initial concentration of effluent (mg/l)            |
|------------------|---|---|
| Q                | = | Flow rate $(m^3/s)$                                 |
| T                | = | Temperature (°C)                                    |
| Co               | = | Effluent concentration at accumulation stage (mg/l) |
| $\tilde{C_2}$    | = | Final concentration of effluent (mg/l)              |
| Cd               | = | Concentration of the decay rate (mg/l)              |
| g                | = | Acceleration due to gravity $(m/s^2)$               |
| ť                | = | Time (second)                                       |
| V                | = | Volume of a tank (m <sup>3</sup> )                  |
| Xo               | = | Initial biomass concentration (mg/l)                |
| X                | = | Final biomass concentration (mg/l)                  |
| S                | = | Substrate concentration (mg/l)                      |
| K                | = | Equilibrium constant (dimensionless)                |
| Qw               | = | Mass flow rate of wastewater (kg/s)                 |
| Xe               | = | Biomass concentration in effluent (mg/l)            |
| Xr               | = | Biomass concentration removed (mg/l)                |
| μ                | = | Viscosity   |
| μ <sub>m</sub>   | = | Specific microbial growth rate (mg/l/s)             |
| Ks               | = | Equilibrium constant of the substrate               |
| So               | = | Initial concentration of the substrate (mg/l)       |
| BOD <sub>5</sub> | = | Biochemical Oxygen Demand (kg/l)                    |
| $\theta_{\rm C}$ | = | Retention time                                      |
| K <sub>d</sub>   | = | Equilibrium constant of the decay rate              |
| Ċ                | = | Concentration of the effluent (mg/l)                |
|                  |   |   |

- A = Filter area  $(m^2)$
- NRe = Renold's number (dimensionless)  $a = Density (kg/m^3)$
- $\rho$  = Density (kg/m<sup>3</sup>) P = Pressure (N/m<sup>2</sup>)
- L = Reactor length (m)
- D = Impeller diameter (m)

#### REFERENCES

- APHA, 1989. American Public Health Association. Public Health and Managed Care Washington D.C. (126 meeting) November 15-19.
- Atlas R.M. 1988. Microbiology: Fundamentals and Applications. 2<sup>nd</sup> edition, Macmillan Publish Company. New York, pp.100-230.
- American Water Works Association, 1990. Water Treatment New York: McGraw-Hill, pp.72-120.
- Bartz, A.E. 1988. Basic Statistical concepts. 3<sup>rd</sup> edition: Macmillan Publish Company, USA, pp.75-91
- Benfled, L.D. and Bandall, C.W. 1980. Biology Process Design for Wastewater Treatment Upper Saddle River, NJ: Prentice Hall, pp.7-118.
- Camp., T.R. and Messerve, R.L. 1974. Water and its impurities 2<sup>nd</sup> edition, Dowden, Hutchison and Rose, Inc. Standsbury, Pennsylvania, pp.346-412.
- Chikere, B.O.D. 2000. Impact of Eleme Petrochemical effluent on microbial ecology of Eleme River. A Ph.D Dissertation, Department of Microbiology University of Port Harcourt, Nigeria, pp.87-115.
- Davis, M.L. and Cornwell, D.A. 1999. Introduction to Environmental Engineering, 3<sup>rd</sup> edition. McGraw-Hill International, New York, pp.201-232.
- Eckenfelder, W.W. 1989. Industrial Water pollution Control 2<sup>nd</sup> edition, McGraw-Hill Publishing Company, New York, pp.52-200.
- Environmental Guidelines, 1991. Standards for petroleum Resources Ministry of petroleum and Mineral Resources, Lagos, pp.6-10.
- Gerard, K. 1999. Environmental Engineering McGraw-Hill, London, pp.206-217
- Handan, S.A. 1976. Industrial Wastewater management Hand Book McGraw-Hill Publishing Company, pp.830-879.
- Hurley, B.S. 1989. Hazardous Waste Future Consideration J. *Iwem* 3, pp.168-173.
- Internet: fiel//A:/petrochemical. Htm. Pollution prevention and Abatement sources World bank, 15<sup>th</sup> September, 2001.
- James, C.S.L.U. 1980. Standard methods for Examination of Water and Wastewater. APHA-ANWA-WPCF 15<sup>th</sup> edition. Company pp.15-50.
- Kemp, D.D. 1998. Global Environmental Issues: A Climatology Approach. London, pp.8-27.
- King, N.J. 1978. Industry and Environment in the oil industrial and microbial ecosystems. Charter K.N.A. and A.J. Somerville (eds) Hegdon. Pp.55-60.
- Klyhn, F.A. 1979. Protein and Oil Recovery from wastewater, Water Services 83: (998), pp.227-250.
- Lynch, J.M. and Poole, N.J. 1999. Microbial Iconology, a Conceptual Approach, pp.60-90.
- Marshal, S., 1973. Pollutant Removal hand Book, Noyes Data Corporation London, pp.136-140.
- Nwaogazie, I.L. 2000. Probability and Statistics for Science and Engineering Practice, 2<sup>nd</sup> edition, Publisher: Konzults, lagos Nigeria, pp.17-20.

- Onuoha, G.U.C. 1994. Water Quality and Phytoplankton Production in the Okiriki Port Harcourt Section of the Bonny Estuary, pp.56-114.
- Pollution prevention and Abatement Guidelines for Petrochemicals Manufacturing Industry. (PPAGPMI 2010) World Bank Publication. Source: Internet: File://petro chemcial.htm.8/7/02).
- Raven, P.H.: Berg, L.R. and Johnson, G.B. 1998. Environment 2<sup>nd</sup> edition, Saunders College publishing new York, pp.110-200.
- Martin, D.P. & Kulathooran, S. 2007. Impact of Tsunami on the bioaccumulation of trace metals on the clan meretrix coasta in vanjur Estuary karaikal, South East Wast of India., *Journal of Pollution Research*, vol.26, no.2, pp.181-182.
- Srinikethan, G.; Shrihari, S. and Pradeepan, V.S. 2007. Treatment of Domestic Wastewater by Association of up flow anaerobic and aerobic biofilters. *Journal of pollution Research*, vol.26, no.2, pp.184-187.
- Khan, A.K.; Shaikh; Zarina, lanjekar and Nayab, 2007. Lipid level in different body parts of the Green Mussesl, Pernaviridis, Exposed to Zinc chloride in summer season. *Journal of pollution Research* vol.26, no.2, pp.219-211.
- Ukpaka, C. Peter, Ogoni, H.A. and Ikenyiri, P.N., 2004. Mathematical Modelling on the Determination of Substrate Concentration Gradient in a Pond with Continuing Effluent Discharge, Nigerian, *Journal of Research and Production* Vol. 5, No.1, pp.27-134.
- Ukpaka, C. Peter, Ogoni, H.A.; Amadi S.A. and Adebayo T.A., 2005. Mathematical Modeling of the Microbial Growth and Decay Rate of Pseudomonas Species on Biodegradation of Bonny Light Crude Oil. *Global Journal of Pure and Applied Sciences*, vol. 11, No. 3, pp. 423-431.
- Ukpaka C.P. 2006. Modeling the microbial thermal Kinetics system in Biodegradation of n-paraffins", *Journal of Modeling, Simulation and Control (AMSE)*, vol. 67, no.1, pp.61-84.
- Ukpaka, C. P; D.M. Ene and K.K. Dune, 2007. Development of Biokinetic Model for phosphorus production from Natural Aquatic Ecosystem polluted with crude oil, *Nigerian*
- Journal of Research and Production, vol.11, no.1, pp.75-90.
- Ukpaka, C. Peter and Musa, A.S. 2009. Evaluation of crude oil degradation in fresh water contaminated site, *The Nigeria Academic Forum: A multi-disciplinary Journal*, vol.16,no.2, pp. 4-13.
- Ukpaka C.P., Ogoni H.A., Gumus R.H. and Farrow S.T. 2008. Biokinetic model for production of ammonia from urea in a batch reactor, *The Nigerian Academic Forum*, vol.15, no.2, pp.8-24.
- Ukpaka, C.P. and Davis, D.D. 2010. Studying the Biodegradation of petroleum Hydrocarbon in Soil using Pseudomonas sp. In Niger Delta Area of Nigeria. *Multidisciplinary Journal of Academic Excellence*, vol.1, no.2, pp.1-15.
- Ukpaka, C.P. 2010. Predictive Techniques to estimate the oxygen utilization by *Pseudomonas* Aeruginosa in petroleum Hydrocarbon in a fluidized bed Reactor. *ICASTOR Journal of Engineering. Accepted for publication February*, 10.
- Ukpaka, C.P. 2010. Development of mathematical model for the prediction of microbial growth rate of Bacteria and fungi in BTX contaminants Degradation in soil Environment (In Press). *Multidisciplinary Journal of Research Development.*

\*\*\*\*\*\*