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RESEARCH ARTICLE

DISTRIBUTIONAL CHARACTERISTICS OF DIMETHYL SULPHIDE (DMS) RELATED TO PHYTOPLANKTON BIOMASS AND NUTRIENT DYNAMICS IN THE COCHIN ESTUARY

Dayala, V. T., Jose Mathew and *Sujatha C. H.

Department of Chemical Oceanography, School of Marine Sciences, Cochin University of Science and Technology, Cochin-16, Kerala, India

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ABSTRACT

Distribution of Dimethyl Sulphide (DMS) was measured in relation with phytoplankton density in the Cochin estuarine system during the year 2010. A total of 120 species of phytoplankton were identified which represents different distinct classes viz: *Bacillariophyceae* (65), *Chlorophyceae* (25), *Dinophyceae* (21), *Cyanophyceae* (6), *Dictyochophyceae* (1), *Chrysophyceae* (1) and *Zygnematophyceae* (1). The phytoplankton identification reveals that Cochin estuary is a diatom dominated estuary. The maximum concentration of diatom species was high in pre monsoon season (av.57693 cell/m³) followed by monsoon (av.45073 cell/m³) and post monsoon (av.40320 cell/m³) whereas dinoflagellates range av.14413 cells/m³ (post monsoon), av.7840 cells/m³ (pre monsoon) and av.4593 cells/m³ (monsoon). Hydrographical parameters and nutrient distribution were also measured to ascertain a relationship with phytoplankton. Chlorophyll *a*, salinity and phosphate exhibit a positive correlation with DMS. The DMS concentration varied from non detectable levels to 19.5 nM in post monsoon, while (0.2 to 1.8 nM) in pre monsoon and (0.2 to 1.1 nM) in monsoon. Elevated levels of DMS were observed in saline stations of the estuary. The data represented above is the first baseline study of DMS in the Cochin Estuarine system.

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INTRODUCTION

Dimethyl sulphide (DMS) is considered to be the most abundant form of volatile organic sulfur. DMS contributes about two-third of global natural sulfur emission to the atmosphere and extends its role in the sulfur cycle and climate (Lovelock *et al.*, 1972., Rodhe, 1999). The natural occurrence of DMS was first discovered by Haas in 1935. The production of DMS by several classes of phytoplankton through biological activity in the aquatic realm was extensively studied (Lovelock *et al.*, 1972., Charlson, 1987). The concentration of DMS in water sample depends on the production by phytoplankton, other microorganisms, bacterial and photochemical consumption (Andreae, 1986), zooplankton grazing (Dacey and Wakeham, 1986) in addition to microbial decomposition of DMSP to DMS (Andreae, 1985). Dimethylsulphonio propionate (DMSP) is the major precursor of DMS, which is synthesized by marine phytoplankton as an internal cell component. DMSP is also considered as a compatible solute involved in osmoprotection and cryoprotection in algae (Stefels, 2000).

Cochin estuary is classified as a tropical dynamic estuary. Although several studies accounting this dynamic behavior of

the estuary, but no database yet published regarding the complex and consumption of biogenic sulfur gas (DMS). This is the first preliminary report on DMS with respect to phytoplankton community. Estuaries are the cradle grounds for flora and fauna, where sundry activities regularly occur and are the most productive of the aquatic ecosystem. These ecosystems are highly vulnerable and easily subjected to stresses induced by environment or human. Several studies have been accomplished in this estuary on various physico-chemical (Sankaranarayanan and Qasim, 1969., Shyanamma and Balakrishnan, 1973) and biological characteristics (Rao *et al.*, 1975., Madhupratap and Haridas, 1975., Qasim, 2003., Martin *et al.* 2008).

The spatial and temporal variability of DMS production is widely studied (Yang *et al.*, 2000a, b., Jiao *et al.*, 2003). But less data base is available from the Indian sector rather than in estuary (Kumar *et al.* 2009, Shenoy *et al.* 2002, Shenoy and Patil 2003). The distribution of DMS in the marine water was influenced by various environmental factors. Salinity is one of the responsible factor for DMS production, in which algal cells produce organic solutes such as quaternary ammonium compounds (Keller *et al.*, 1999a, b) and tertiary sulfonium compounds (DMSP) (Blunden and Gordon, 1986., Bisson and Kirst, 1995). Previous work by Sunda and Hardison, in 2007 highlights the effect of nitrogen limitation on cellular DMSP and DMS release in marine phytoplankton.

*Corresponding author: Sujatha C. H.

Department of Chemical Oceanography, School of Marine Sciences, Cochin University of Science and Technology, Cochin-16, Kerala, India.

MATERIALS AND METHODS

Site details

Cochin estuary is a bar-built micro-tidal system connected to the Arabian Sea at two locations one at Cochin (latitude 9°10' N) and at Azhikode (latitude 10°10' N). The estuary is flanked between two parts: the southern arm extending from Cochin to the south and the northern arm extending from Cochin to Azhikode. The Cochin bar mouth is about 450 m wide, whereas the Azhikode inlet is relatively narrow. The Cochin metropolis receives an annual rainfall of 320 cm, of which 60% occurs during the southwest monsoon period, July–Sept (Qasim 2003). The estuary, receives a high volume of fresh water annually ($20 \times 10^9 \text{ m}^3 \text{ year}^{-1}$) from the six rivers in the State, Kerala (Srinivas *et al.*, 2003). During the months of December to April, construction of a salinity barrier bund at Thanneermukkam virtually cuts off the tidal propagation further towards south and modifies the circulation patterns in the remaining part of the estuary. The samples were collected during 2010 in three prominent seasons; post monsoon (Jan.), pre monsoon (Apr.) and monsoon (Aug.). Fifteen stations in the estuary were selected for DMS measurements. Sampling was conducted twice in each season and the average values for each parameter is reported. The sampling sites were best ascribed in Figure 1 and the specifications are as follows.

features leading to a hodgepodge of multidimensional behaviors. The remaining stations from 11 to 15 flows closely through industrial region and many small and large scale industries on the river bank discharges effluents directly into water ultimately leading varying amount of nutrients in to the lower river.

Sampling and analysis

Samples have been taken for qualitative and quantitative analysis of physico-chemical parameters. Surface and bottom water samples were collected by using a clean plastic bucket and Niskin water sampler respectively. The temperature was measured by using a thermometer. Salinity was calculated by Mohr-Knudsen titration technique. Water samples were analyzed for nutrients (nitrate, phosphate and silicate) within 6 hours after collection following standard procedures and protocols (Grasshoff *et al.*, 1999).

Chlorophyll *a* analysis

Chlorophyll *a*, in water samples were determined by filtering the sample through GF/C filter paper and extracting with 90% acetone (Parsons *et al.*, 1984). The mixture is kept for overnight under dark condition. After incubation, the mixture is grinded well and centrifuged at 5000rpm for 15 minutes. The supernatant was used for the pigment analysis using UV visible spectrophotometer (GENESYS 10UV).

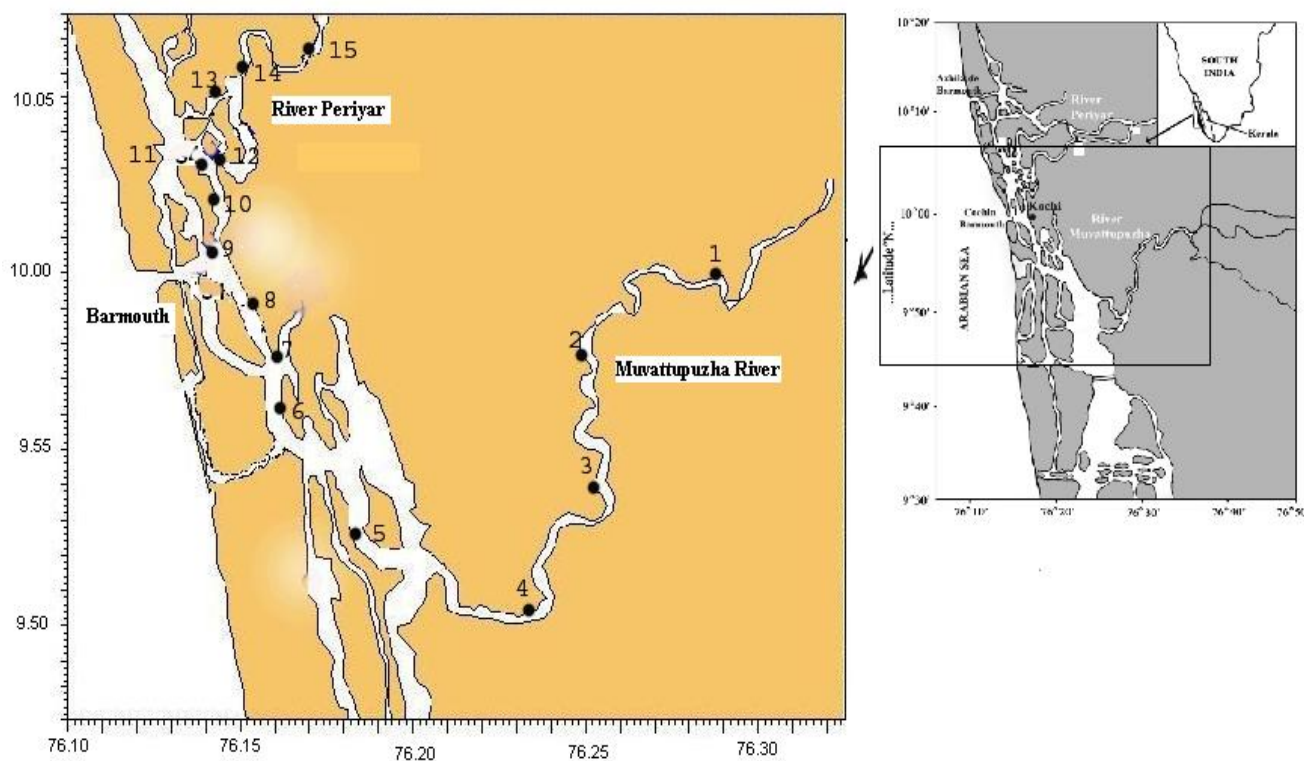


Fig.1. Study area in the Cochin estuary showing the station locations

The stations numbered from 1 to 4 are generally fresh water in nature and industrial effluents are relatively less than municipal waste. The stations 5 to 7 lie in the coastline section. Moving together, stations 8 to 10 become estuarine in character and its connection to the Arabian Sea exhibits vivid

Analysis and identification of phytoplankton

For analyzing phytoplankton cell counts and composition, water samples were filtered through a phytoplankton net of 20 μ mesh size made of bolting silk. The filtrate was preserved in 3% Lugol's iodine solution. A setting and siphoning

procedure was followed to concentrate samples from 250ml to 20ml (Utermohl, 1958). For counting phytoplankton cells and identification of genera and species, the concentrated samples were thoroughly shaken and from each, 1ml replicates were transferred into a sedge wick-rafter plankton counting chamber and examined by using biological microscope (OLYMPUS; MLX) at 200x magnification. The planktonic micro algae filtered from 100 L of water was made up to a fixed volume concentrate. 1 ml of this sample was transferred to the sedge wick-Rafter counting cell (the volume of this chamber is 1 ml). The number of micro algae present in the cell 1000 grids was calculated. Repeated the counting for three times and took the average. The total number of planktonic algal species present in water sample was calculated using the formula,

$$N = \frac{n * v}{V}$$

N= total number of phytoplankton cell per liter of water filtered; n= average number of phytoplankton cells in 1 ml of plankton sample; v = volume of plankton concentrate (ml); V =volume of total water filtered (L).

DMS analysis

Water samples for DMS were transferred to 60 ml amber colored bottles. Care was taken to avoid atmospheric contact and samples were preserved immediately in the dark at 4°C. Analysis was completed within ten hours. DMS was measured using AGILENT 7890 gas chromatograph equipped with flame photometric detector (FPD). A known volume of sample (10ml) was purged (15min) using nitrogen gas and the stripped sulfur gases were passed through moisture traps (ice bath, glass wool and potassium carbonate). These traps were replaced very frequently. The sulfur gases were cryogenically (liquid nitrogen) trapped in a teflon loop. The loop was then transferred to a water bath, maintained at >80°C, for removal of the trapped gases. Separation was done on a DB-5 capillary column. Temperature ramp program was set at initial 80°C for 5 minutes and final 180°C for 25 minutes. DMS calibrations were done using DMS standard (Sigma), ethanol (Merck) and milli-Q water. The retention time of DMS was 2.8min and detection limit 0.05 nM. The linear detection range is from 0.2 nM to 25 nM. The calibration curve with precision of analysis was presented in Figure 2.

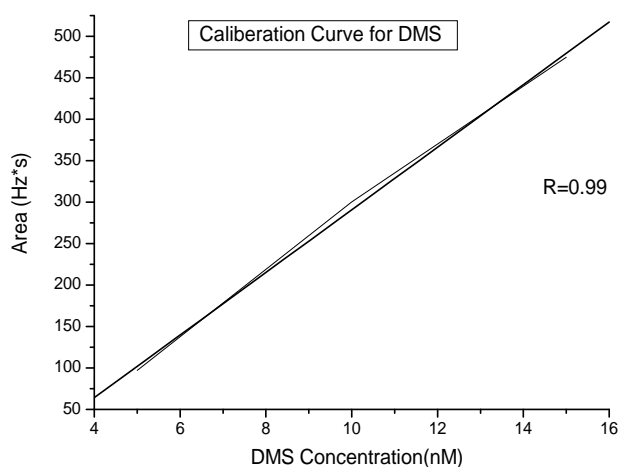


Fig. 2. Calibration curve of Dimethyl sulphide (DMS)

RESULTS

Hydrographical parameters and nutrient distribution

Hydrographical parameters serve as nucleus for investigating water quality. Estuaries which are in the brim of oceanic and marine environment undergo rapid changes which are reflected in the quality of water. Sea water intrusion and fresh water mixing pose serious fluctuations in the estuarine ecosystem and as a result hydrodynamic parameters keeps on oscillating. During the study, temperature varied from 32-34.5°C pre monsoon (PRM), 26-34°C post monsoon (POM) and 26-30°C monsoon (MON), where as in bottom it ranges from 31-33°C (PRM), 29-31°C (POM) and 26-27.5°C (MON). The average temperature recorded in the surface was pre monsoon>post monsoon>monsoon with 33°C, 31°C and 29°C and the bottom also follows the same trend with 32°C, 30°C, 26.6°C respectively (Figure 3).

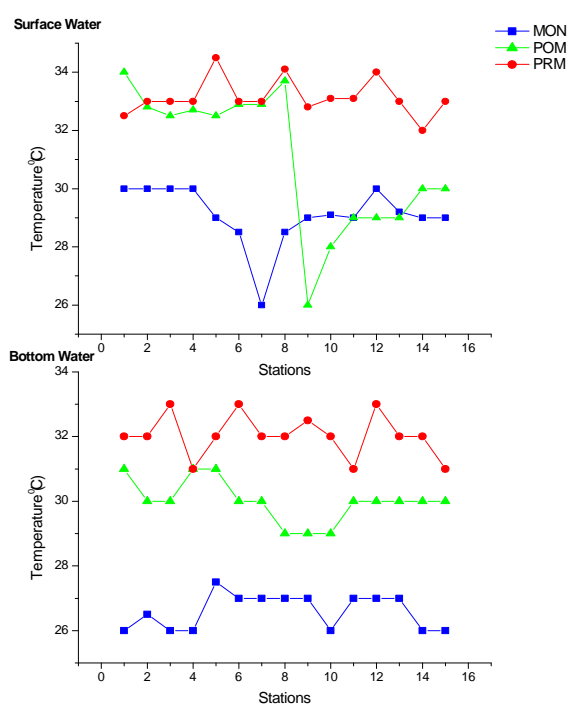


Fig.3. Distribution pattern of temperature at both surface and bottom waters in three prominent seasons

Levels of nitrate ranges from 3.03-16.74 $\mu\text{mol L}^{-1}$ (MON), 3.7-23.75 $\mu\text{mol L}^{-1}$ (POM) and 1.03-30.99 $\mu\text{mol L}^{-1}$ (PRM) in the surface. The bottom values vary 7.55-38.44 $\mu\text{mol L}^{-1}$ (MON), 2.56-15.72 $\mu\text{mol L}^{-1}$ (POM) and 5.01-26.15 $\mu\text{mol L}^{-1}$ (PRM). The phosphate values showed considerable discrepancies in all seasons and in surface ranges from 0.88-6.56 $\mu\text{mol L}^{-1}$ (MON), 0.83-8.42 $\mu\text{mol L}^{-1}$ (POM) and 1.08-8.12 $\mu\text{mol L}^{-1}$ (PRM). In bottom it fluctuates from 2.15-8.22 $\mu\text{mol L}^{-1}$ (MON), 1.32-7.88 $\mu\text{mol L}^{-1}$ (POM) and 0.54-10.23 $\mu\text{mol L}^{-1}$ (PRM). The surface silicate concentrate varied between 0.06-5.82 $\mu\text{mol L}^{-1}$ (MON), 21.67-92.4 $\mu\text{mol L}^{-1}$ (POM), 8.52-82.16 $\mu\text{mol L}^{-1}$ (PRM) and the bottom between 0.48-4.36 $\mu\text{mol L}^{-1}$ (MON), 27.09-130 $\mu\text{mol L}^{-1}$ (POM) and 9.51-78 $\mu\text{mol L}^{-1}$ (PRM) (Figure 4).

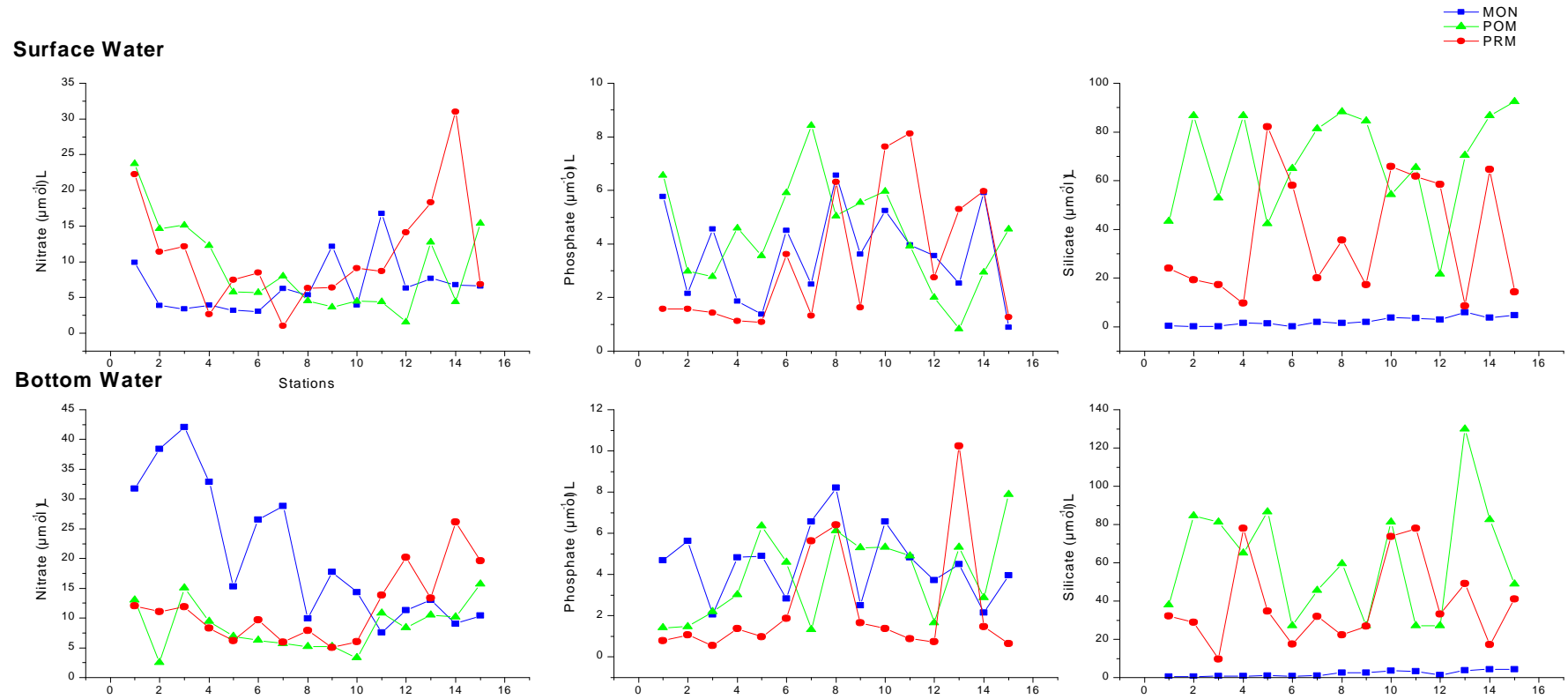
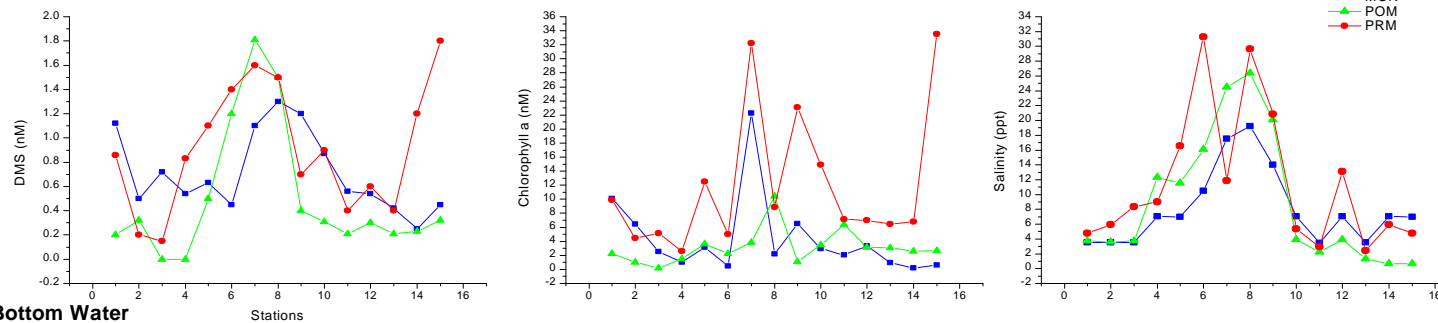


Fig. 4. Distribution pattern of nutrients (Nitrate, Phosphate and Silicate) at both surface and bottom waters in three prominent seasons

Table 1. Comparison of DMS values with other estuaries

Estuary	Concentration Range (nM)	Average Concentration (nM)	References
Estuary in North America	1-18	-	Iverson et al. 1989
Canal de Mira	0-18	2.9-5.3	Cerqueira & Pio, 1999
Scheldt Estuary	0-2.5	0.4-0.6	Scaire et al. 2002
Zuari Estuary	0.3- 15.4	-	Shenoy et al. 2002
Gironde Estuary	0-1.7	0.2-0.7	Min Hu et al. 2005
Elbe Estuary	0-2.5	0.9	Min Hu et al. 2005
Rhine Estuary	0-10	0.2	Min Hu et al. 2005
Loire Estuary	0.5-3.6	1.3	Min Hu et al. 2005
Pearl River Estuary	0.05-56.7	3.0-8.6	Min Hu et al. 2005
Cochin Estuary	0-19.5	-	This Work

Surface Water



Bottom Water

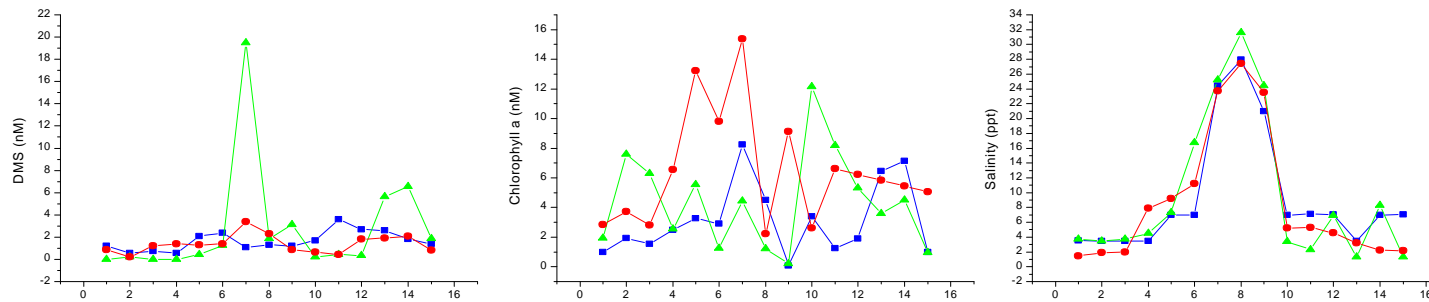


Fig. 5. Distribution pattern of DMS, Chl.a and Salinity at both surface and bottom waters in three prominent seasons

Allocation of DMS, Chlorophyll a and Salinity

The DMS values in the surface estuary ranged from 0.25-1.3nM (MON), 0.2-1.8nM (PRM) and 0-1.81nM (POM) and bottom DMS concentration varied from undetectable levels to 19.5nM (POM), 0.2-3.4nM (PRM) and finally 0.54-3.6nM in (MON). Higher concentration of DMS was observed in the post monsoon season. The elevated concentrations of DMS were observed in the saline regions of the estuary especially station 6, 7, 8 and 9. Comparison of DMS concentration with other estuaries is appended in (Table 1) and the values are in agreement with that of Iverson *et al.* (1989) and Cerqueira and Pio, 1999. Chlorophyll *a*, a significant biomarker for assessing phytoplankton biomass was spatially and temporally estimated. The surface concentrations ranges from 0.23-10.44nM (POM), 4.49-33.52nM (PRM) and 0.22-22.24nM (MON) where as in bottom it varies from 0.22-12.14nM (POM), 2.21-15.36nM (PRM) and 0.07-8.24nM (MON). Salinity the foremost key of an estuary in the surface varies in MON (3.48-19.25ppt) where as considerable increase in PRM (2.41-31.26ppt), POM (0.65-26.4ppt) and in bottom (3.49-27.96ppt) in MON, (1.29-31.57ppt) POM, and (1.45-27.45ppt) PRM respectively (Figure 5).

Total phytoplankton biomass

In the present study, a total of 120 species of planktonic microalge were identified within 7 classes viz. *bacillariophyceae*, *chlorophyceae*, *dinophyceae*, *chrysophyceae*, *cyanophyceae*, *dictyochophyceae* and *zygnematophyceae*. Qualitative and quantitative analysis of planktonic microalgae reveals *bacillariophyceae* is the dominant taxa with 65 species and the abundance consist of (av.57693 cell/m³) PRM, (av.45073 cell/m³) MON and (av.40320 cell/m³) POM, followed by *chlorophyceae* with 25 species, *dinophyceae* with 21 species; *dictyochophyceae*, *cyanophyceae*, *chrysophyceae* and *zygnematophyceae* with comparatively low numbers (Table 2 & Figure 6). The dominant diatom species comprised of *Skeletonema costatum*, *Coscinodiscus spp.*, *Thalassiothrix spp.*, *Nitzschia spp.*, *Chaetoceros spp.* and *Rhizosolenia spp.* whereas *Ceratium spp.*, *Dinophysis spp.*, *Diplosalis spp.*, *Protoperidinium spp.* and *Prorocentrum spp.* were the dominant dinoflagellates.

Statistical analysis

A statistical analysis were also employed to find the correlation between the different environmental parameters such as temperature, salinity, nitrate, phosphate, silicate, chlorophyll a and DMS using SPSS 13.

DISCUSSION

The phytoplanktons are the major source of DMS production in the marine environment (Kiene *et al.*, 1996). Earlier, Aneeshkumar and Sujatha, 2012 reported that in Cochin estuarine system, fucoxanthin was the most abundant carotenoid pigment which indicates profuse of diatom community. Regional studies in the Coast of Goa (Shenoy *et al.*, 2012), North Sea (Turner *et al.*, 1988) and the East Coast of the U.M.A. (Iverson *et al.*, 1989) have accounted the measurement of regional DMS fluxes and compared these with biological parameters such as phytoplankton biomass and

Table 2. Qualitative identification of phytoplankton records

CHLOROPHYCEAE (25)	
<i>Ankistrodesmus falcons</i>	<i>Hemidiscus hardmannianus</i>
<i>Arthodesmus convergens</i>	<i>Hyalodiscus subtilis</i>
<i>Chlorella sp.</i>	<i>Leptocylindrus danicus</i>
<i>Chlorococcum sp.</i>	<i>Navicula henneidyi</i>
<i>Closterium sp.</i>	<i>Nitzschia closterium</i>
<i>Coelastrum sp.</i>	<i>Nitzschia fasciculata</i>
<i>Euastrum sp.</i>	<i>Nitzschia longissima</i>
<i>Micrasterias foliacea</i>	<i>Nitzschia marina</i>
<i>Pediastrum duplex</i>	<i>Nitzschia seriata</i>
<i>Pediastrum simples</i>	<i>Nitzschia sigma</i>
<i>Pleodorina sp.</i>	<i>Pleurosigma directum</i>
<i>Scenedesmus arcuatus</i>	<i>Pseudonitzschia seriata</i>
<i>Scenedesmus quadricauda</i>	<i>Rhizosolenia imbricata</i>
<i>Selenastrum gracile</i>	<i>Rhizosolenia robusta</i>
<i>Sphaeroszoma granulatum</i>	<i>Rhizosolenia styliiformis</i>
<i>Staurastrum asteroideum</i>	<i>Skeletonema costatum</i>
<i>Staurastrum gracile</i>	<i>Surirella elegans</i>
<i>Staurastrum leptocladium</i>	<i>Surirella sp.</i>
<i>Staurastrum pingue</i>	<i>Thalassionema nitzschioides</i>
<i>Staurastrum sp.</i>	<i>Thalassiosira subtilis</i>
<i>Tetraedron trigonum</i>	<i>Thalassiothrix frauenfeldii</i>
<i>Tetraspora sp.</i>	<i>Thalassiothrix longissima</i>
<i>Ulothrix tenuissima Kuetzing</i>	<i>Triceratium affine</i>
<i>Volvox aureus Ehrenberg</i>	<i>Triceratium favus</i>
<i>Xanthidium antilopaenum</i>	<i>Triceratium reticulam</i>
BACILLARIOPHYCEAE (65)	<i>Triceratium sp.</i>
<i>Actinocyclus sp.</i>	<i>Tropidoneis sp.</i>
<i>Achnanthes sp.</i>	DINOPHYCEAE (21)
<i>Amphiprora alata</i>	<i>Ceratium breve</i>
<i>Amphora sp.</i>	<i>Ceratium furca</i>
<i>Asterionella Formosa</i>	<i>Ceratium lineatum</i>
<i>Asterionella japonica</i>	<i>Ceratium macroceros</i>
<i>Asteromphalus flabellatus</i>	<i>Ceratium tripos</i>
<i>Aulacoseira granulate</i>	<i>Dinophysis caudata</i>
<i>Bacillaria paradoxa</i>	<i>Dinophysis miles</i>
<i>Bacteriastrium varians</i>	<i>Diplopsalis lenticula</i>
<i>Biddulphia aurita</i>	<i>Diplopsalis acuta</i>
<i>Biddulphia mobilianis</i>	<i>Gonyalux sp.</i>
<i>Biddulphia rhombus</i>	<i>Gymnodinium sp.</i>
<i>Biddulphia sp.</i>	<i>Heterocapsia sp.</i>
<i>Cerataulina bergonii</i>	<i>Noctiluca miliaris</i>
<i>Cerataulina pelagic</i>	<i>Peridinium claudicans</i>
<i>Chaetoceros affinis</i>	<i>Prorocentrum maximum</i>
<i>Chaetoceros coarctatus</i>	<i>Prorocentrum micans</i>
<i>Chaetoceros decipiens</i>	<i>Protoperidinium depressum</i>
<i>Chaetoceros denticulatum</i>	<i>Protoperidinium oceanic</i>
<i>Cheatoceros densus</i>	<i>Protoperidinium pellucidum</i>
<i>Coscinodiscus asteromphalus</i>	<i>Protoperidinium sp.</i>
<i>Coscinodiscus centralis</i>	<i>Scrippsella sp.</i>
<i>Coscinodiscus granii</i>	CYANOPHYCEAE (8)
<i>Coscinodiscus marginatus</i>	<i>Anabaena sp.</i>
<i>Coscinodiscus oculis-iridis</i>	<i>Katagnymene spiralis</i>
<i>Coscinodiscus perforatus</i>	<i>Merismopedia sp.</i>
<i>Coscinodiscus radiates</i>	<i>Nostoc colony</i>
<i>Coscinodiscus subtilis</i>	<i>Tolypothrix sp.</i>
<i>Cyclotella sp.</i>	<i>Trichodesmium sp.</i>
<i>Cyclotella meneghiana</i>	CHRYSOPHYCEAE (1)
<i>Cyclotella striata</i>	<i>Dinobryon sp.</i>
<i>Cylindrotheca closteridium</i>	DICTYOCHOPHYCEAE (1)
<i>Cymbella marina</i>	<i>Dictyocha fibula</i>
<i>Dityllum brightwelli</i>	ZYGNEMATOPHYCEAE (1)
<i>Dityllum sol</i>	<i>Spirogyra sp.</i>
<i>Fragilariopsis sp.</i>	
<i>Gyrosigma sp.</i>	

chlorophyll *a*. In the present study a significant correlations were observed on DMS with chlorophyll *a* (Table 3) in the surface waters of monsoon, post monsoon and pre monsoon season. Similar trends were cited by (Barnard *et al.*, 1982., Tanzer, 1992 and Belviso *et al.*, 1993a). Pingree *et al.*, 1975 reported the high concentration of DMS in surface water and showed a clear association with chlorophyll *a* levels.

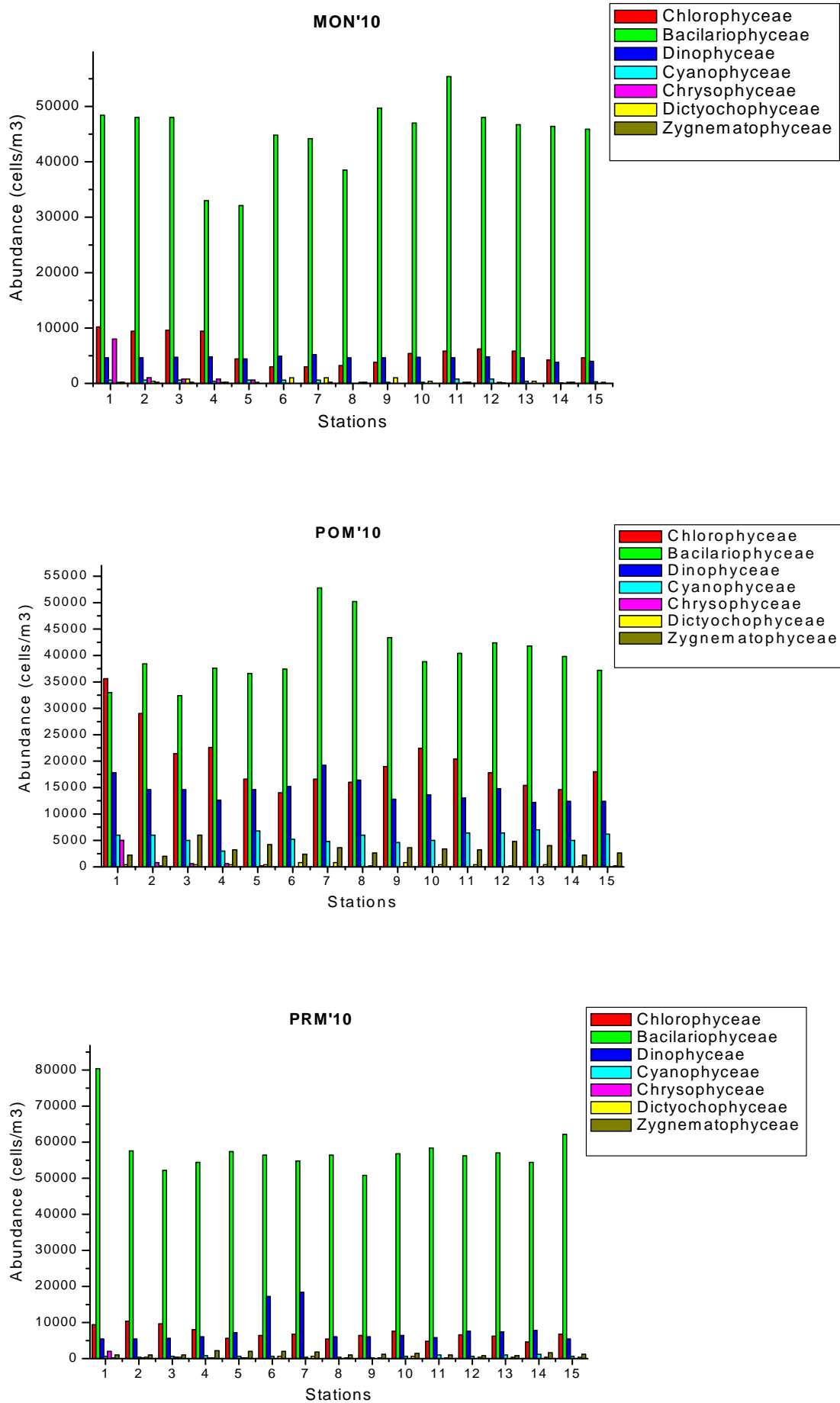


Fig. 6. Abundance of phytoplankton species (cells/m³) at three prominent seasons during the study period

The physiological factors such as light intensity and salinity greatly affect the amount of DMS from DMSP produced (Dickson and Krist, 1986., Van Bergeijk *et al.*, 2002). Laboratory studies have shown that intracellular DMSP in marine macroalgae (Dickson *et al.*, 1980., 1982., Reed, 1983) and in phytoplankton (Vairavamurthy *et al.*, 1985) increases with increasing salinity. Similar result was cited by Zhang *et al.* in (1999). The present study also supports this statement. Salinity favorably affected the DMS production in surface samples of both POM and MON season whereas bottom sample resulted in PRM season. Furthermore, modeling studies also reflects that limited nutrient concentrations favors an increase in DMS concentrations apart from the increase in salinity, chlorophyll *a*, temperature and light (Laroche *et al.*, 1999). Inadequate positive correlation between DMS and phosphate were observed both in surface (POM) and bottom (PRM) waters during the study period whereas a strong negative correlation were exhibited between DMS and nitrate in (MON) bottom water (Table 3).

DMS concentrations were determined in surface waters at stations 6,7,8,9,11,13,14 and 15 inclusive of riverine, estuarine and coastal system in all the seasons. However, the maximum concentration of DMS (19.5 nM) was found at station 7 during post monsoon season at bottom water. Impact of stratification may be the reason for higher salinity which inturn leads to high DMS values (Ramamirtham *et al.*, 1986) and enrichment of DMS producing phytoplankton species. The major identified DMS producing species include: *Skeletonema costatum*, *Cylindrotheca closterium*, *Thalassiosira spp.*, *Rhizosolenia spp.*, *Heterocapsa spp.*, *Prorocentrum minimum*, *Prorocentrum micans* and *Scrippsiella spp.* This observation well in support with the results of previous studies conducted by Keller *et al.* in 1989, and investigated the DMS production in some strains of marine phytoplankton such as dinoflagellates (*Ceratium spp.*, *Heterocapsa spp.*, *Prorocentrum spp.* and *Scrippsiella spp.*), and diatoms (*Skeletonema costatum*, *Thalassiosira spp.*, *Rhizosolenia spp.*, *Cylindrotheca closterium* and *Nitzschia spp.*). Besides, the

Table 3. Correlation analysis of DMS with other environmental variables at three different seasons in both surface and bottom waters

MON-Surface	Variables	DMS	Chl. a	Salinity	NO ₃	PO ₄	SiO ₃	Temp.
	DMS	1	.560(*)	.626(*)	.180	.365	-.324	-.324
	Chl. a		1	.401	.101	-.089	-.262	-.617(*)
	Salinity			1	-.082	.188	-.126	-.713(**)
	NO ₃				1	.135	.338	-.040
	PO ₄					1	-.174	.046
	SiO ₃						1	-.116
	Temp.							1
MON-Bottom	DMS	1	.061	-.141	-.682(**)	-.108	.389	.508
	Chl. a		1	.297	-.173	.276	.233	.160
	Salinity			1	-.239	.440	.056	.427
	NO ₃				1	-.112	-.779(**)	-.321
	PO ₄					1	-.038	.192
	SiO ₃						1	-.151
	Temp.							1
POM-Surface	DMS	1	.523(*)	.806(**)	-.316	.594(*)	.237	.365
	Chl. a		1	.393	-.402	.128	.098	.120
	Salinity			1	-.343	.598(*)	.289	.270
	NO ₃				1	.038	.060	.469
	PO ₄					1	.187	.250
	SiO ₃						1	.018
	Temp.							1
POM-Bottom	DMS	1	-.115	.457	-.149	-.235	.027	-.103
	Chl. a		1	-.454	-.259	-.185	.311	-.119
	Salinity			1	-.492	.064	-.360	-.484
	NO ₃				1	.052	-.004	.426
	PO ₄					1	.102	-.306
	SiO ₃						1	.078
	Temp.							1
PRM-Surface	DMS	1	.594(*)	.411	-.232	-.052	.131	.078
	Chl. a		1	-.024	-.393	-.276	-.216	-.044
	Salinity			1	-.371	-.066	.199	.428
	NO ₃				1	.255	.183	-.479
	PO ₄					1	.459	-.058
	SiO ₃						1	.344
	Temp.							1
PRM-Bottom	DMS	1	.447	.517(*)	.023	.594(*)	-.307	.198
	Chl. a		1	.428	-.311	.112	-.062	.054
	Salinity			1	-.576(*)	.389	-.192	.118
	NO ₃				1	-.178	-.173	-.059
	PO ₄					1	-.016	-.005
	SiO ₃						1	-.691(**)
	Temp.							1

species composition of phytoplankton is also a determining factor for the DMS production in an aquatic system (Groene, 1992) and our studies revealed that *Skeletonema costatum*, an ubiquitous species enriched in all the seasons at saline stations.

Conclusion

The present research work provides baseline information on the distributional characteristics of the DMS related hydrography, nutrients, biomass and taxonomic composition of the phytoplankton. Microscopic observation of phytoplankton cell counts points that in general the diatom community dominated and the abundant groups in terms of species diversity and density rather than other taxonomic groups. Results obtained in this study suggest that the production of DMS was species specific and influenced by different growth stages of algae. Moreover the salinity conditions also displayed the physiological and ecological complexity of the DMS production. There was no methodical drift in the DMS values; yet reviewing of this all pervading gas becomes unique in nature due to its intervention with global climate.

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